


**SAFETY DATA  
SHEET**

**Material Name: Compressed Gas , N.O.S  
( Argon 1-98%/ Carbon Dioxide 2-99%)**

| <b>Section 1 – Product and Company Identification</b> |  |
|---|--|
| Product Identifier:                                   | Compressed Gas N.O.S   |
| Other means of identification:                        | Mix Gas, weld gas  |
| Product Uses:   | Industrial manufacturing and laboratory including inerting, steel making, metals processing, welding, etc. |
| Supplier Details:                                     | Western Gasco Cylinders Ltd.<br>2169 Peardonville Road<br>Abbotsford BC<br>V2T 6J7                         |
| Emergency Phone Number:                               | (613) 996-6666   |

| <b>Section 2 – Hazards Identification</b>                     |   |
|---|---|
| Classification in accordance with paragraph (d) of §1910.1200 | Gas Under Pressure – Compressed gas Simple asphyxiant   |
| Signal word<br>Hazard statement(s)<br><br>Symbol              | Warning<br>Gas in pipelines may be under pressure, cylinders may explode if heated May displace oxygen and cause rapid suffocation<br><br>   |
| Precautionary statement                                       | Read completely and follow all Safety Data Sheets before use<br>Do not handle until all safety precautions have been read and understood<br>Never enter an area where nitrogen may have caused an oxygen deficiency<br>Close valve after each use and when empty<br>Use a backflow preventative device in piping<br>Use equipment and materials rated for cylinder pressure Use and store only outdoors or in well- ventilated area Protect from sunlight |

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|                                  |   |
|----------------------------------|---|
| Hazards not otherwise classified | None  |
| Toxicity                         | Non-toxic but may displace oxygen which can cause dizziness, unconsciousness and death by asphyxiation. |

| <b>Section 3 – Compositions / Information of Ingredients</b> |   |
|--|---|
| Chemical Name & Formula                                      | Ar/CO2                                  |
| Common Name and Synonyms                                     | Compressed Gas N.O.S                    |
| CAS Number   | 7440-37-1/124-38-9                      |
| Purity   | Argon 1% – 99%<br>Carbon Dioxide 2%-99% |

| <b>Section 4 – First Aid Measures</b>               |   |
|---|---|
| Inhalation  | Simple asphyxiant, may cause acute effects including dizziness, drowsiness, nausea, rapid breathing, unconsciousness, and death. Victim may not be aware of asphyxiation.<br>Immediately remove victim to fresh air containing sufficient oxygen.<br>If not breathing provide artificial respiration or oxygen by trained personnel, get immediate medical attention.<br>Rescuers must not enter an oxygen deficient area without self contained breathing apparatus. |
| Skin Contact  | No adverse effects expected. Very cold gas may cause frostbite.   |
| Eye Contact   | No adverse effects normally expected from gas. Avoid high pressure or very cold gas. Remove contact lenses.<br>Flush with water, seek medical attention if irritation persists.   |
| Ingestion   | Not an expected route of exposure, refer to inhalation section above.   |
| Most important symptoms, effects, acute and delayed | Refer to asphyxiation acute effects as per inhalation above   |

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| Immediate medical attention and special treatment needed | If symptoms occur, seek medical advice and attention. |
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| <b>Section 5 – Fire Fighting Measures</b>                                 |  |
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| Suitable extinguishing media  | Compressed Gas N.O.S is not flammable, will not burn.<br>Use appropriate extinguishing media for surrounding fire.   |
| Special hazards arising (e.g. nature of any hazardous combustion process) | Compressed Gas N.O.S is a non-flammable gas.<br>Heat from fire may cause pressure to rise and container to burst.<br>Cool any containers with water if possible.   |
| Special protective equipment and precautions for firefighters             | Wear appropriate protective gear and self-contained breathing apparatus.<br>Never attempt to rescue a suspected asphyxiation victim without proper precautions, training and equipment to also avoid exposure to oxygen deficient conditions.<br>Argon gas is heavier than air at same temperature which can cause it to concentrate in low areas and lead to oxygen deficiency. |

| <b>Section 6 – Accidental Release Measures</b>                   |  |
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| Personal precautions, protective equipment, emergency procedures | First responders should ensure oxygen concentration in area is safe (>19.5%) or be trained and use self-contained breathing apparatus before attempting to rescue a victim.<br>Evacuate personnel to safe area, do not allow personnel to walk or drive in area that is potentially oxygen deficient.<br>Use oxygen monitors to ensure adequate oxygen levels.<br>Never enter suspected oxygen deficient area without being properly trained and wearing a self-contained breathing apparatus.<br>Argon gas is heavier than air at same temperature which can cause it to concentrate in low areas and lead to oxygen deficiency.<br>Prevent spreading of vapors through sewers, ventilation systems and confined areas. |
| Methods and materials for containment and clean up               | Isolate any leaking sources of argon if it can be done safely. Ventilate the area if possible.   |
| <b>Section 7 – Handling and Storage</b>                          |  |

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|   |   |
|---|---|
| <p>Precautions for safe handling</p>                                | <p>Protect system components against physical damage.<br/>           Use adequate ventilation.<br/>           Avoid inhalation and potential confined space areas, use oxygen monitors where appropriate.<br/>           Never work on a pressurized system.<br/>           Wear gloves when moving cylinders.<br/>           Safety glasses always recommended when working with compressed gases. Refer to CGA Safety Bulletin SB-2 "Oxygen Deficient Atmospheres" for additional recommendations.</p>  |
| <p>Conditions for safe storage, including any incompatibilities</p> | <p>Use storage containers, piping, valves and fittings designed for storage and distribution of gaseous argon.<br/>           Argon gas is heavier than air at same temperature which can cause it to concentrate in low areas and lead to oxygen deficiency.<br/>           Protect cylinders against physical damage. Store in cool, dry, well-ventilated, fireproof area, away from flammable materials and corrosive atmospheres. Store away from heat and ignition sources and out of direct sunlight. Do not store near elevators, corridors or loading docks. Do not allow area where cylinders are stored to exceed 52°C (125°F).<br/>           Move cylinders with a suitable hand-truck. Do not drag, slide or roll cylinders. Do not drop cylinders or permit them to strike each other. Secure cylinders firmly. Leave the valve protection cap in-place (where provided) until cylinder is placed into service and after it is taken out of service.<br/>           Use designated CGA fittings and other support equipment. Do not heat cylinder by any means to increase the discharge rate of the product from the cylinder. Use check valve or trap in discharge line to prevent hazardous backflow into the cylinder. Do not use oils or grease on gas-handling fittings or equipment.</p> |

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| <b>Section 8 – Exposure Controls / Personal Protection</b>     |   |
|--|---|
| Permissible exposure limits                                    | There are no exposure limits for this product.<br>Oxygen levels should be kept above 19.5% for all personnel.   |
| Appropriate Engineering Controls                               | Adequate ventilation.<br>Low Oxygen monitors and alarms in areas where oxygen deficiency is possible.<br>Pressurized systems to have relief valves properly sized, calibrated and vented.   |
| Individual protection measures / personal protective equipment | Use self-contained breathing apparatus for entering any suspected oxygen deficient area.<br>Use personnel oxygen monitors.<br>Gloves and safety shoes for handling containers/cylinders.<br>Safety glasses / face protection if exposure to discharged gases, eye wash station.<br>Check systems regularly for leaks. |

| <b>Chemical Name</b>       | <b>ACGIH TLV</b>                         | <b>OSHA PEL</b>  | <b>NIOSH IDLH</b>   |
|----------------------------|--|--|---|
| ARGON<br>7440-37-1         | : See Appendix F: Minimal Oxygen Content | None   | None  |
| CARBON DIOXIDE<br>124-38-9 | STEL: 30000 ppm<br>TWA: 5000 ppm         | TWA: 5000 ppm<br>TWA: 9000 mg/m <sup>3</sup><br>(vacated) TWA: 10000 ppm<br>(vacated) TWA: 18000 mg/m <sup>3</sup><br>(vacated) STEL: 30000 ppm<br>(vacated) STEL: 54000 mg/m <sup>3</sup> | IDLH: 40000 ppm<br>TWA: 5000 ppm<br>TWA: 9000 mg/m <sup>3</sup><br>STEL: 30000 ppm<br>STEL: 54000 mg/m <sup>3</sup> |

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| <b>Section 9 – Physical and Chemical Properties</b> |                 |   |  |
|---|-----------------|---|--|
| <b>Property</b>                                     | <b>Value</b>    | <b>Property</b>                         | <b>Value</b>   |
| Appearance  | Colorless       | Upper/Lower Explosive Limit             | NA   |
| Odor  | Odorless        | Vapor Pressure                          | NA   |
| Odor Threshold                                      | NA              | Vapor Density                           | 0.103 lb/ft <sup>3</sup> @ 70°F<br>1.65 kg/m <sup>3</sup> @ 21.1°C   |
| Molecular Weight                                    | 39.95 g/mol     | Specific Volume                         | 9.67 ft <sup>3</sup> /lb @ 70°F<br>0.61 m <sup>3</sup> /kg @ 21.1 °C |
| pH  | NA              | Relative Density to Air                 | 1.38   |
| Melting / Freezing Point                            | -308°F / -189°C | Solubility                              | Slight in water  |
| Boiling Point                                       | -302°F / -186°C | Partition Coefficient: noctanol / water | NA   |
| Flash Point   | NA              | Auto Ignition Temperature               | NA   |
| Evaporation Rate                                    | NA              | Decomposition Temperature               | NA   |
| Flammability  | Non-flammable   | Viscosity (dynamic)                     | 0.0226 centipoise @70 °F   |

**Section 10 Stability and Reactivity**

|                                    |   |
|------------------------------------|---|
| Reactivity                         | Not reactive under normal conditions        |
| Chemical Stability                 | Stable at normal temperatures and pressures |
| Possibility of Hazardous Reactions | None  |

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|                                  |   |
|----------------------------------|---|
| Conditions to Avoid              | Exposure to reactive metals at high temperatures<br>High concentrations causing oxygen deficiency atmosphere leading to asphyxiation effects (see sections 4, 6, 7 & 8) |
| Incompatible Materials           | None known  |
| Hazardous Decomposition Products | None  |

| <b>Section 11 Toxicology Information</b>                              |  |
|---|--|
| Information on likely routes of exposure                              | No chemical toxicity<br>Inhalation – simple asphyxiant<br>Ingestion – not an expected route<br>Skin – no effects expected normally, cold gas may cause frostbite<br>Eye – no effects expected normally, cold gas may cause frostbite |
| Symptoms related to physical, chemical, toxicological characteristics | As a simple asphyxiant, the presence of high concentrations causing an oxygen deficiency in air has symptoms which include dizziness, drowsiness, nausea, unconsciousness, and death.  |
| Delayed, Immediate, chronic effects from short and long term exposure | As a simple asphyxiant, the immediate effects of high concentrations causing oxygen deficiency in air include dizziness, drowsiness, nausea, unconsciousness, and death.   |
| Numerical measures of toxicity  | LD50 – not available LC50<br>– not available   |
| Carcinogen Listing  | Not carcinogenic   |


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| <b>Section 12 – Ecological Information</b> |   |
|--|---|
| Ecotoxicity                                | None  |
| Persistence and degradability              | Not applicable. Normal air is approximately 0.9% argon by volume. |
| Bio-accumulative potential                 | No information available  |
| Mobility in Soil                           | No information available  |
| Other Adverse effects                      | No known other effects  |

**Section 13 Disposal Considerations**

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|--|--|
| Waste residues and disposal guidelines | <p>Product will normally dissipate in air, however argon gas is heavier than air at same temperature which can cause it to concentrate in low areas and lead to oxygen deficiency.</p> <p>Dispose of any contents or containers in accordance with applicable regulations. Cylinders should be returned in original shipping container/method with any valves closed and protective plugs or caps securely in place.</p> |
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| <b>Section 14 – Transport Information</b> |   |
|---|---|
| US DOT UN ID Number                       | UN1956  |
| UN Proper Shipping Name                   | Compressed Gas, N.O.S   |
| DOT Transportation Hazard Class           | DOT Class 2.2<br>(Non-Flammable compressed gas)<br>Emergency Response Guide 121  |
| Packing Group                             | Not Applicable  |
| Environmental Hazards                     | None  |

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|----------------------|--|
| Transport Bulk Codes | Not Applicable   |
| Special Precautions  | <p>Ensure vehicle driver is aware of the potential hazards of the load and knows what to do in the event of an accident or an emergency.</p> <p>Because gas is denser than air, it can cause oxygen deficiency in low areas, with gas spreading along the ground, isolate area to avoid personnel exposure or other vehicles entering the area.</p> <p>High pressure gas cylinders should have outlet valves closed, with plugs/valve caps secured in place.</p> <p>Load space must be separated from driver compartment.</p> <p>Cylinders should be firmly secured from moving or falling during transport.</p> |

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| <b>Section 15 - Regulatory Information</b>  |
| <p>US Federal TSCA Toxic Substance Control Act - exempted</p> <p>US EPA SARA Title III Section 312 hazard Category: Sudden release of pressure hazard US</p> <p>States Right-To-Know Lists: Massachusetts, New Jersey, Pennsylvania</p> |

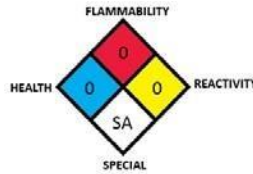
**Section 16 Other Information**

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US Nation Fire Protection Agency (NFPA) hazard ratings:

(Scale of 0 to 4, with 0 = lowest increasing to 4 = highest hazard, refer to NFPA for details related to the relative rating for each category)

Health: 0 Fire: 0  
 Reactivity: 0  
 Special: SA (Simple Asphyxiant)



US Hazardous Material Information System (HMIS) ratings:

(Scale: 0 = minimal, 1 = slight, 2 = moderate, 3 = serious, 4 = severe)

|                  |   |
|------------------|---|
| HEALTH           | 0 |
| FLAMMABILITY     | 0 |
| PHYSICAL HAZARDS | 3 |

New SDS: 29 June 2018 Rev 0

**USE OF THIS INFORMATION:**

Western Gasco Cylinders Inc. offers this information to promote the safe use of this product through awareness of hazards and safety information. Those who use or transport or sell this product to others should: 1) Disseminate this information internally to all workplace areas, employees, agents and contractors likely to encounter this product

- 2) Provide supplemental hazards awareness, safety information, operation and maintenance procedures to the workplace areas and employees, agents and contractors likely to encounter this product
- 3) Furnish this information to all their customers who purchase this product
- 4) Ask each purchaser or user of the product to notify its employees and customers of the product hazards and safety information.

**DISCLAIMER OF EXPRESSED AND IMPLIED WARRANTIES:**

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